

# Optimal operation of a Petlyuk Distillation Column: Energy Savings by Over-fractionation



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## Abstract

This work shows the unexpected result that over-fractionating one of the product streams in a Petlyuk distillation column may be optimal from an energy point of view. Analytic expressions for the potential energy savings are derived using the Underwood equations. The energy savings by over-fractionation may be further increased by bypassing some of the feed and mixing it with the over-fractionated product to meet product specifications. Normally, the energy savings are small, so the main significance of our results is to point out that over-fractionating is optimal in some cases.

## 1 Introduction

- The Petlyuk distillation column, see Figure 1(a), with a pre-fractionator ( $C_1$ ) and a main column ( $C_{21}$  and  $C_{22}$ ), is an interesting alternative to the conventional cascade of binary columns for separation of ternary mixtures. The potential savings are reported to be of approximately 30% in both energy and capital cost [4].
- It is well known that if the products have different economic value, it may be economically optimal to over-fractionate the low value product in order to produce more of the more valuable products.
- Here we intend to show that we in fact can **save energy** by over-fractionating one of the product streams.
- It is known from literature that for a conventional binary distillation column, bypassing a portion of the feed to the products does not affect the energy demand to produce the specified products [2].

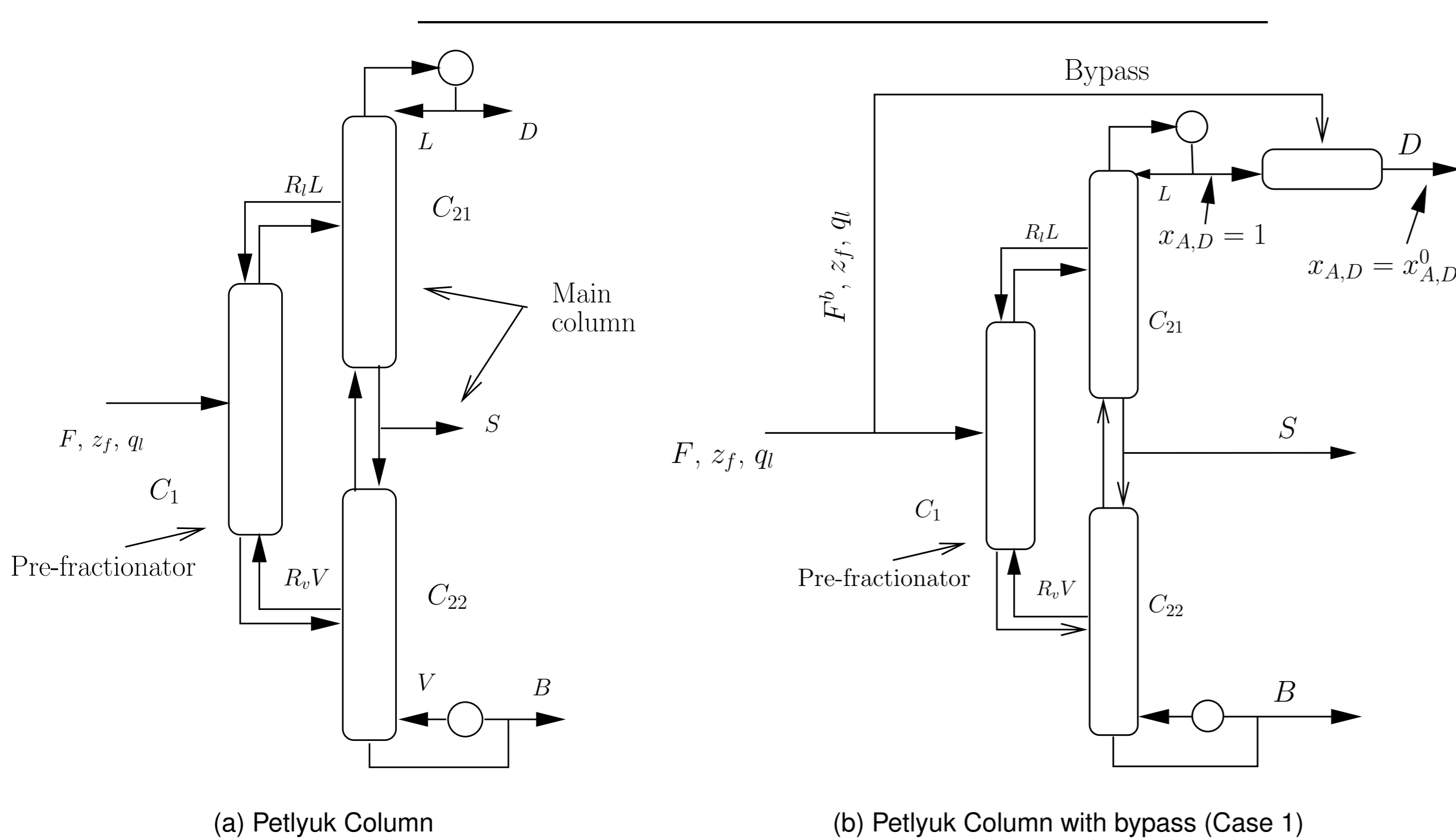


Figure 1: Sketch of the Petlyuk column without (a) and with bypass (b)

## 2 System description

- Feed  $F$**  consist of 3 components ( $A, B$  and  $C$ ):
  - Composition:  $\mathbf{z}_F = [z_A \ z_B \ z_C]^T$
  - Liquid fraction:  $q_f$
  - Relative volatility:  $\alpha = [\alpha_A \ \alpha_B \ \alpha_C]^T$
- Products**
  - Distillate**, flow  $D$ ,  $x_D = [x_{A,D} \ x_{B,D} \ x_{C,D}]^T$
  - Side-stream**, flow  $S$ ,  $x_S = [x_{A,S} \ x_{B,S} \ x_{C,S}]^T$
  - Bottom-stream**, flow  $B$ ,  $x_B = [x_{A,B} \ x_{B,B} \ x_{C,B}]^T$
- Operational objective:** Minimize energy consumption (minimize boilup ( $V$ )) with given minimum purity:

$$\min_{\mathbf{u}} V(\mathbf{u}) \quad (1)$$

$$x_{A,D} \geq x_{A,D}^0, \quad x_{B,S} \geq x_{B,S}^0, \quad x_{C,B} \geq x_{C,B}^0 \quad (2)$$

- $\mathbf{u} = [L \ V \ S \ R_L \ R_V]^T$  is the vector of steady-state degrees of freedom (manipulated inputs).
- $x_{i,j}^0$  is the minimum fraction of the main component  $i \in \{A, B, C\}$  in each product stream  $j \in \{D, S, B\}$ .

## 3 $V_{min}$ -diagram and Underwood equations

- The  $V_{min}$ -diagram, see Figure 3(a), is a graphical representation of the energy requirements in distillation columns [3] and is based on the Underwood equations [5].
- Assumptions: (1): Constant molar flows. (2): Constant relative volatility. (3): Infinite number of stages.
- For a three-product column it can be shown that the minimum energy diagram for the Petlyuk column with sharp splits maps the  $V_{min}$  diagram for the pre-fractionator  $C_1$  operated at the preferred split [3].
- Same diagram applies for non-sharp splits, and the minimum boilup is given by eq. (3)

$$V_{T,min}^{Petl} = \max(V_{T,min}^{D/S,B}, V_{T,min}^{DS/B}) = \max(V_{T,min}^{C_{21}}, V_{B,min}^{C_{22}} + (1 - q_f)F) \quad (3)$$

where

$$V_{B,min}^{C_{22}} = -B \left[ \frac{\alpha_A(1 - x_{C,B})}{\alpha_A - \theta_B} + \frac{\alpha_B(x_{C,B})}{\alpha_B - \theta_B} \right] \quad V_{T,min}^{C_{21}} = D \left[ \frac{\alpha_A x_{A,D}}{\alpha_A - \theta_A} + \frac{\alpha_B(1 - x_{A,D})}{\alpha_B - \theta_A} \right] \quad (4)$$

where  $\theta_A = \theta_A(\mathbf{z}_F, q_f, \alpha)$  and  $\theta_B = \theta_B(\mathbf{z}_F, q_f, \alpha)$  are the Underwood roots carried over from  $C_1$  to  $C_{21}$  and  $C_{22}$  respectively.

## References

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- Three cases of operation [3]:

- **Case 1:**  $C_{22}$  is limiting: Separation  $B/C$  is the most difficult separation (peak  $C_{22}$  is above peak  $C_{21}$ ).
- **Case 3:**  $C_{21}$  is limiting: Separation  $A/B$  is the most difficult separation, as illustrated in Figure 3(a).
- **Case 2:** Balanced main column: Required vapor load are equal.

- Important:**

- **Case 1** with  $x_{C,B}$  constant: Minimum boilup proportional to  $B$ .
- **Case 3** with  $x_{A,D}$  constant: Minimum boilup proportional to  $D$

## 4 Energy savings by over-fractionation

- Based on the material balance of the column, explicit expressions for  $B$  and  $D$  are derived [1].
- Case 1 (Case3):** Energy savings ( $E_S = \frac{V^0 - V}{V^0}$ ) when increasing the purity from  $x_{A,D}^0$  ( $x_{C,B}^0$ ) to  $x_{A,D}$  ( $x_{C,B}$ )

$$\text{Case 1: } E_S^{C_{22}} = \frac{\frac{1}{x_{A,D}^0} - \frac{1}{x_{A,D}}}{\frac{z_C}{z_A} \frac{1}{x_{C,S}^0} + \frac{1}{x_{A,D}^0} - \frac{1}{z_A}} \quad \text{Case 3: } E_S^{C_{21}} = \frac{\frac{1}{x_{C,B}^0} - \frac{1}{x_{C,B}}}{\frac{z_A}{z_C} \frac{1}{x_{A,S}^0} + \frac{1}{x_{C,B}^0} - \frac{1}{z_C}} \quad (5)$$

- Approximate savings without bypass:

- **Case 1:**  $E_S \approx x_{C,S}^0 x_{B,D}^0$     **Case 2:**  $E_S \approx x_{C,S}^0 x_{B,B}^0$

- Physical explanation (Case 1):**

Energy savings is possible since (1) by over-fractionating in the top component  $B$  is moved from the distillate to the side-stream, see Figure 2 for an illustration. (2) Without violating the constraint in the side-stream, component  $C$  may now be moved from the bottom stream to the side-stream. (3) Since the boilup is proportional to the amount of bottom product eq.(4), the energy input is reduced.

- Physical explanation (Case 3):**

Same as Case 1, but now with over-fractionating in the bottom.

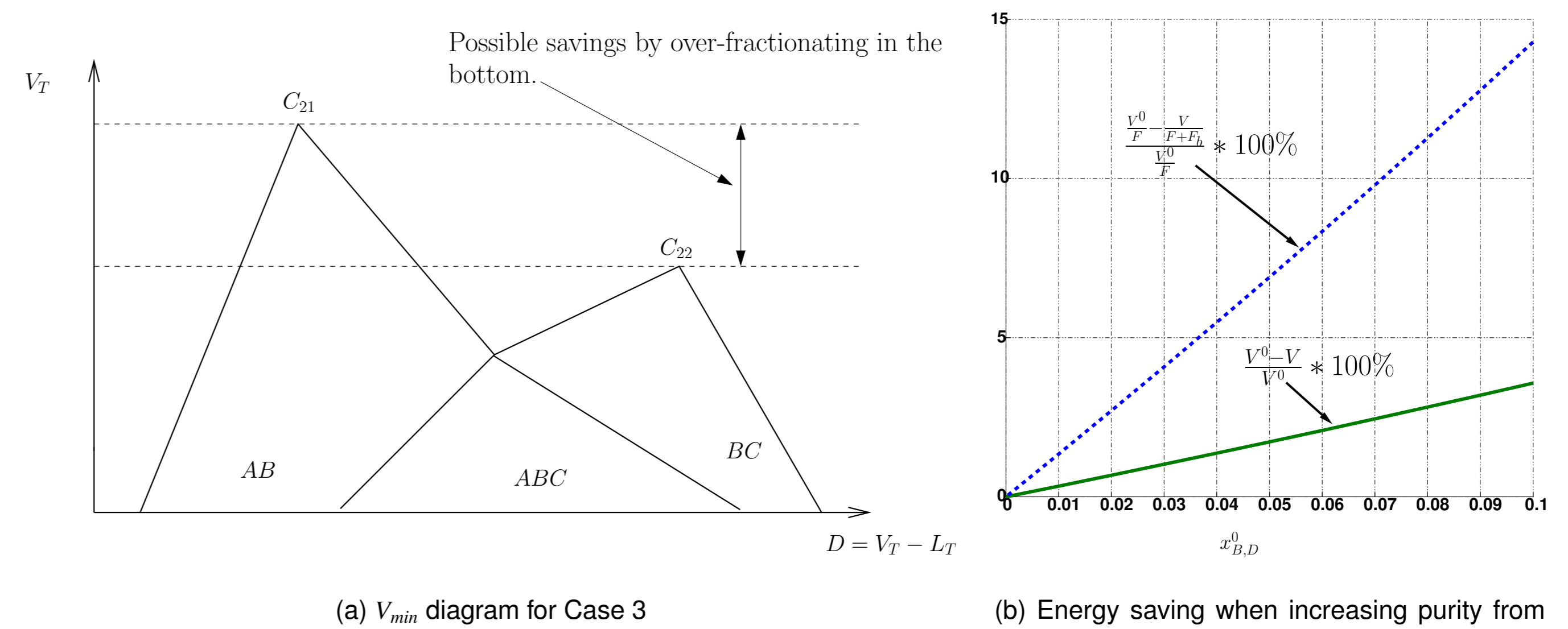
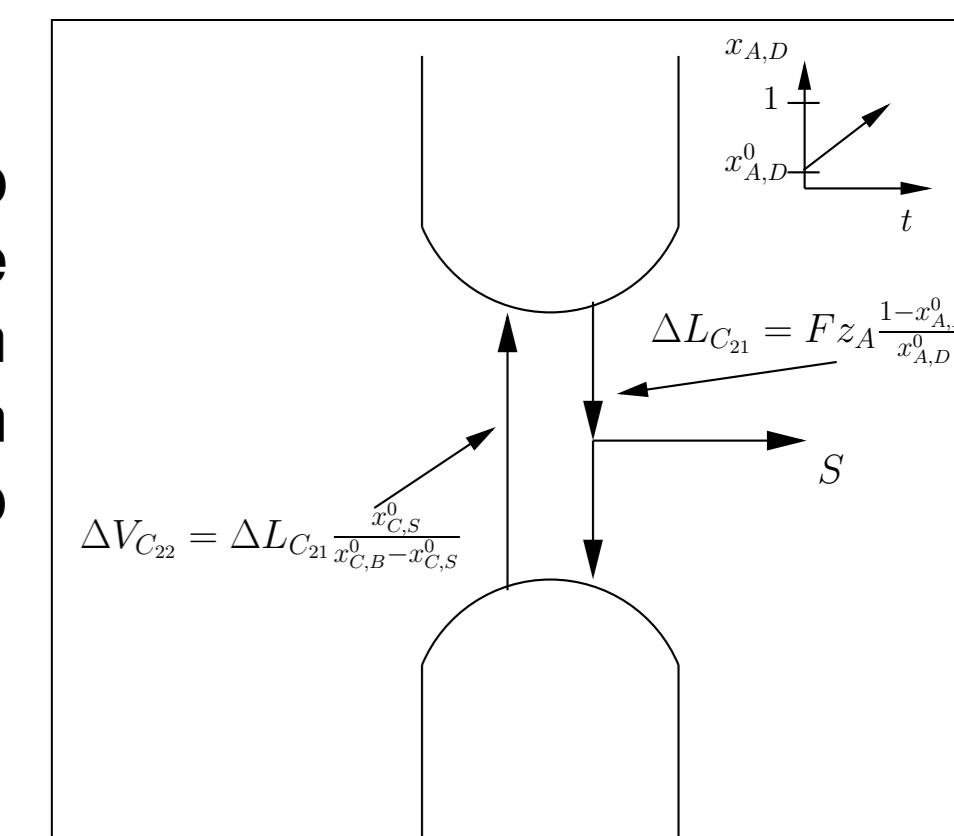


Figure 3:  $V_{min}$  and energy savings for for Case 3.

## 5 Additional savings using bypass

- Over-fractionating one of the products makes it possible to bypass some of the feed to the product to fulfill the composition constraint, reducing the energy input further, see Figure 1(b).
- Amount of bypass when over-fractionating to pure products ( $x_{A,D} = 1$  or  $x_{C,B} = 1$ ):

$$\text{Case 1: } F_B^{C_{22}} = D \frac{x_{B,D}^0}{1 - x_{B,D}^0 - z_A} \quad \text{Case 3: } F_B^{C_{21}} = B \frac{x_{B,B}^0}{1 - x_{B,B}^0 - z_C}$$

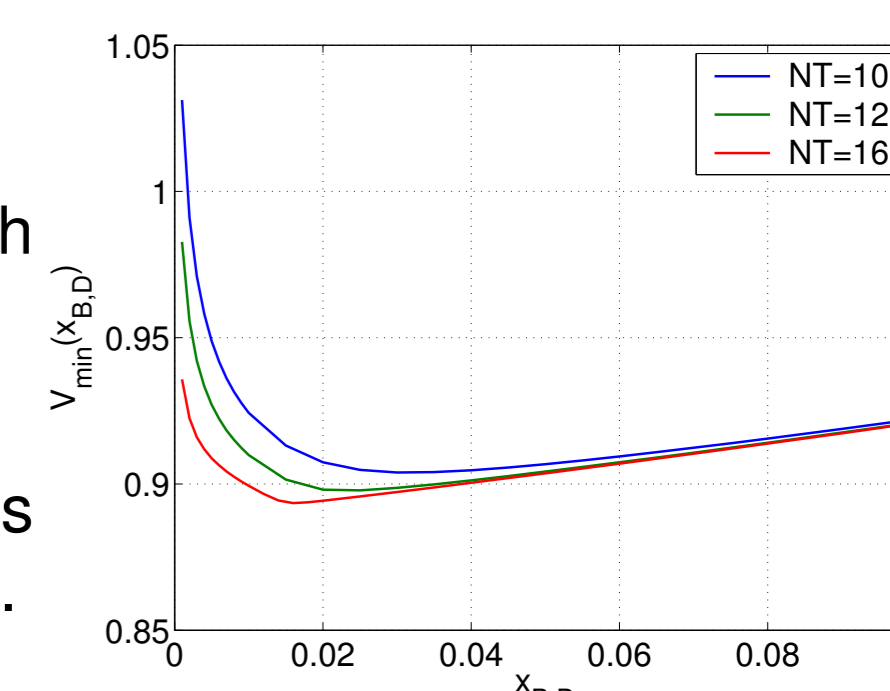
- But:** Introduces a component ( $C$  or  $A$ ) into the product ( $D$  or  $B$ ) that normally is not present
- Figure 3(b) illustrate the potential savings, for a specific case. Up to 4% energy savings without bypass and 13% energy savings with bypass.

## 6 Confirmation of results for finite number of stages

- Simulations carried out to verify the results. Assumptions model:

- Constant relative volatility. Finite, equal number of stages in each section. Constant molar flows.
- $z_F = [0.5 \ 0.3 \ 0.2]$ ,  $\alpha = [9 \ 3 \ 1]$ ,  $q_f = 1$ ,  $x_{B,S}^0 = 0.9$ ,  $x_{C,B}^0 = 0.97$

- Simulation confirms that one may save energy when the column has sufficient number of stages. Results also confirmed using HYSYS®.



## Conclusions

- Optimal from an energy point of view to over-fractionate one of the streams in the Petlyuk distillation column.
- Additional savings possible if bypassing some of the feed to the over-fractionated product.
- Explicit expressions for the achievable energy savings derived based on the Underwood equations assuming infinite number of stages.
- Energy savings possible due to different vapor load demands in the two main column sections.
- Results have been confirmed for finite number of stages.